Name (Print)	Department
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## Chemistry 685-Safety Seminar Hazardous Waste Disposal and Spill Control January 4, 2001

## Circle the ONE Correct Answer For Each Question

## **Hazardous Waste**

- 1) Laboratory conditions that allow chemical "unknowns" to be created include:
  - a. Improper storage methods
  - b. Students and supervisors not performing periodic inspections of chemicals
  - c. Students not labeling containers
  - d. All of the above
- 2) When is it appropriate to use chemical abbreviations and structures on the yellow Hazardous Waste tags found on the red safety solvent cans?
  - a. Never
  - b. When it is convenient
  - c. When adding the same solvent over and over again
  - d. Only for high molecular-weight compounds
- 3) Hazardous wastes are defined by the **EPA** as being either specifically "listed" or they exhibit a "characteristic" of being hazardous. Which of the items below are the correct hazardous "characteristics"?
  - a. Flammability, corrosivity, toxicity (heavy metals), and reactivity
  - b. Liquids, solids, and gases
  - c. Radioactive and Infectious
  - d. Anything that smells bad
- 4) Of the materials listed below, which waste materials are inappropriate for the red safety (solvent) cans?
  - a. Mercury
  - b. Silica Gel
  - c. Concentrated acids
  - d. All of the above

- 5) The main purpose of the flash arrestor in a safety can is:
  - a. To filter out solid materials
  - b. To neutralize acids as the liquid enters the can
  - c. To act as a "heat sink" and prevent the propagation of flame into the can
  - d. To balance the can to make it easier to pour
- 6) Glassware and broken glassware require special disposal procedures because:
  - a. Glassware can be contaminated with hazardous materials
  - b. Broken glass can be a puncture or cutting hazard
  - c. All of the above
- 7) "Sharps" that have not been in contact with human fluids can be disposed of by:
  - a. Throwing them into the regular trash
  - b. Placing them with glassware trash
  - c. Placing them in a rigid container and then placing them in the glass disposal boxes

## **Spill Control**

- 8) The two primary methods of spill control include:
  - a. Funneling and channeling
  - b. Containment and evaporation
  - c. Adsorption and neutralization
  - d. Dilution and evaporation
- 9) For small, low hazard spills you should:
  - a. Be prepared to adsorb or neutralize the spill yourself
  - b. Call 911
  - c. Pour it down the drain
  - d. Let it evaporate up the fume hood
- 10) For larger, more hazardous spills you should:
  - Call the Safety Coordinator or Environmental Health and Safety (EHS)
  - b. Contain the spill if you can and warn others about the spill
  - c. Block the area to prevent people from entering the lab
  - d. All of the above
- 11) Once a spill is cleaned up, the spilled materials should be:
  - a. Put into an appropriate container and labeled
  - b. Left for the custodians to clean up
  - c. Thrown into the trash
  - Poured down the drain